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game, then you can find all of our cracked software here. Catalytic, acidic
phosphorus species have been widely studied to the extent that several

mechanisms have been proposed and utilized for various commercial applications.

For example, it is known that acid-treated and calcined phosphorus powder catalyzes the reaction of NaOH with K_2S_5 , to form $K_2P_3O_{10}$.

The resulting potassium tetrasulfide and triphosphorus pentasulfide are useful in lubricant compositions such as greases and extreme pressure engine oils. It is known that tributoxyphosphine sulfide and triphenoxyphosphine sulfide react with metal borides to form hydride and oxyphosphide complexes. Thus, when tributoxyphosphine sulfide is heated with K_3P in the presence of CuB_2 , the following products are formed:

$Cu_2B_3P_5$; $PA1 K_3P_3S_4$; and $PA1$

(tributoxyphosphine sulfide). B_3P_5 . $PA1$ 1. Thermal decomposition of $[P(HS)_2Ph]_{1-2}$ ($P(HS)_2Ph$ is

1,3,4,6,7,10-hexakis(diphenylphosphino)decane) yields

1,4-bis(diphenylphosphino)anthrene. ##STR1## $PA1$ 2. The product of the reaction of 1,3,4,6,7,10-hexakis(diphenylphosphino)decane with PCl_3 is

1,4-bis(diphenylphosphino)anthrene. ##STR2## Thermal treatment of $[(EtO)_2C_6H_3][P(HS)_2Ph]_{sup}$